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I mentioned a few weeks ago that I started a course in organic chemistry, but the life of a student is far from dull. In fact, sometimes it's so exciting you can hardly stand it. That was my experience this week, as I had my first science lab today and the lesson involved mixing solutions with k rama rao inorganic chemistry pdf download. If you've been following me for some time now then you know that the reason I have been taking notes on Neil Gaiman is because he wrote about ichthyosaurs before they were scientifically classified as dinosaurs- ichthyosaurs are one of those reasons why I believe science is magic. He also wrote about that time he tried to dig a hole to China with his friends, but I don't understand the science behind why the hole didn't end up in China. At any rate, k rama rao inorganic chemistry pdf download. The lab today was quite different from last week's lab, where we were working with hydrochloric acid and trying to get it out of a beaker before it ate through its glass prison. This week we were doing the practical for our theory lesson on solution concentration, which is all about how much of each component there is in a solution. The different components were: One of the most common ways we learn chemistry is by doing reactions. This week we had to make a mixture and see which two components reacted with each other, and we had to mix them in the correct sequence. It turns out it's hard to mix different solutions together, as they tend to separate if you don't mix them just right. However, I do know that it is absolutely imperative that you bring your solutions into contact with each other as quickly as possible, or else they separate. When this happens you have what's called a phase-separation equilibrium. K rama rao inorganic chemistry pdf download. This week, we had to do this lab in our practicals. We were given three different solutions, each of which was in a separate beaker. The student in front of me used one beaker to mix all three solutions together, then placed it in the freezer to cool the mixture. It took about an hour for the solution to cool, but when I opened the freezer it was bubbling away in all directions! K rama rao inorganic chemistry pdf download. The other student placed each solution in turn into another beaker which contained the third known solution. Then they stood there stirring the contents of the two beakers until they could see that they were mixing. They went through this process three times, and then placed it in the freezer to cool. K rama rao inorganic chemistry pdf download. The first student returned to his bench to find that his mixture had started boiling, and there was steam shooting up all over his bench! He ran back into the classroom, where one of our teachers completely freaked out (he's like that), hit the fire alarm, and sent every single one of us outside.

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